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FERTILIZER MANUAL

IFDC

1998

In formulating liquid mixed fertilizers, it is not sufficient to know that all ingredients are water soluble. Many reactions may occur that result in formation of water insoluble compounds of micronutrients.

Clear liquids used to blend with:

NPK

ATS
Ammonium Chloride
Anhydrous Ammonia (under closed loop conditions)
Aqua Ammonia
Urea Solution
UAN
AN
Nitrate
APP
Ortho Phosphate
KTS
K Hydroxide
K Acetate
K Carbonate
KCL
CTS
CA Chloride
CAN

MICROS

Zinc CL
Zinc SO4
Zinc EDTA
Zinc CL+ EDTA
Zn SO4 + EDTA
Zinc NO3
Zinc SO4 + CA or Gluco or Lignin or HA or FA etc.
MN + all the above
Fe + all the above
Cu + all the above
Boron + monoethanolamine
Boron + HA or FA

Compatibility of various ingredients

CALCIUM + ATS No Makes CASO₄ (Gypsum)
+ KTS No Makes CASO₄ (Gypsum)

Exception is a reacted CTS but it will form precipitate over time.

+ APP No Makes Calcium Phosphate (an insoluble compound)

ORTHO Phosphate requires a fully chelated Micronutrient

Poly phosphate can tolerate most clear complexes CA and Gluco but not lignin

REDLINE™

6-12-2 With Soygreen® Technology

GUARANTEED ANALYSIS

Total Nitrogen (N).....	6%	Copper (Cu).....	0.05%
Available Phosphate (P ₂ O ₅).....	12%	0.05% Chelated Copper (Cu)	
Soluble Potash (K ₂ O).....	2%	Iron (Fe).....	0.40%
		0.40% Chelated Iron (Fe)	
		Manganese (Mn).....	0.05%
		0.05% Chelated Manganese (Mn)	
		Zinc (Zn).....	1.0%
		1.0% Chelated Zinc (Zn)	

Derived from urea, anhydrous ammonia, ammonium nitrate, monammonium phosphate, diammonium phosphate, monopotassium phosphate, dipotassium phosphate, copper EDTA, ORTHO-ORTHO iron EDDHA, manganese EDTA, and zinc EDTA.

GENERAL RECOMMENDATIONS

Redline® contains many nutrients that are necessary for plant growth as well as the same technology that is used in Soygreen® that enhances plant growth and development. Redline® may be used as a soil or foliar application to any food or fiber crop where the addition of one or more of the nutrients contained in Redline® would be beneficial.

When used along with sound agronomic practices, Redline® will enhance plant growth and increase early season root development to maximize yield potential.

Redline® can enhance the yield potential of many crops including corn, dry edible beans, potatoes, sugarbeets, sunflowers, and wheat.

DIRECTIONS FOR USE

SOIL

Apply 1 to 5 gallons per acre as a starter fertilizer or as a soil applied broadcast application in a balanced fertility program. Water may be added to increase application rate.

POSTEMERGENCE

Apply 1 to 3 gallons per acre to an actively growing crop to supplement an existing fertilizer program or to correct an existing nutrient deficiency. Postemergence applications should be made in a minimum of 10 gallons per acre of water.

3 gallons of Redline® contains approximately 1 lbs. of Soygreen®.

PRECAUTIONARY STATEMENTS

CAUTION: Redline® is considered to be a low health hazard and a non-inhalation hazard. Always follow good industrial hygiene practices. Harmful if swallowed. If conscious, immediately give large quantities of water and induce vomiting. Seek medical attention immediately. Avoid contact with skin. In the event of skin contact, flush immediately and thoroughly flush with water. Causes eye irritation. Flush eyes immediately and thoroughly with water for 15 minutes. If irritation persists, seek medical attention.

ENVIRONMENTAL HAZARD

In the case of a spill, contain spill and maximize recovery.

CONDITIONS OF SALE AND WARRANTIES

Seller warrants that the product conforms to its chemical description and is reasonably fit for the purpose stated on the label only when used in accordance with label directions under normal conditions of use. SELLER MAKES NO OTHER EXPRESS OR IMPLIED WARRANTIES EITHER OF MERCHANTABILITY OR FITNESS FOR A PARTICULAR USE. IN NO CASE WILL SELLER BE HELD LIABLE FOR CONSEQUENTIAL, SPECIAL OR INDIRECT DAMAGES RESULTING FROM THE HANDLING, STORAGE, OR INCORRECT USE OF THIS PRODUCT.

NET CONTENTS: 250 GALLONS
10.50 lbs per Gallon

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CAUTION
KEEP OUT OF REACH OF CHILDREN
STOP
READ ENTIRE LABEL BEFORE USING

GREEN Ammonium Polyphosphate



Potassium Thiosulfate



Zinc Ligno-sulfonate representative of MN CU etc.





CU Chelate

ZN Chelate

Zinc Complex

Ammoniated

MN Chelate

Chelation

The objectives of the chelation process in soils are:

- To increase the availability of the nutrient
- To prevent mineral nutrients from forming insoluble precipitates
- To reduce toxicity of some metal ions to plants
- To increase the mobility of plant nutrients

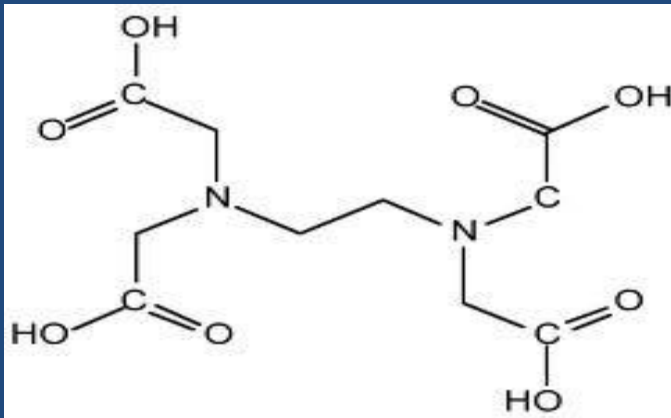
The objective of Chelation in liquid fertilizers is primarily for compatibility.

Prevent the formation of insoluble compounds

Also to prevent salt damage in the case of sulfates.

EDTA Structure and Sample

EDTA's are clear in color, water soluble and protect the associated metal from reacting with other ingredients. They are expensive per unit of nutrient however will mix with almost anything, including orthophosphates. Most effective in the soil.



Orthophosphate

- Orthophosphate is derived from White Acid that is typically 75% acid solution with an analysis of 0-55-0.
- The product is typically used in industrial applications and is free of impurities and is often referred as food grade acid.
- When used for fertilizer, it is reacted with ammonia, urea Liquor, or potassium hydroxide to neutralize the acid. This will make commonly known blends of 0-20-20, 9-18-9, or 3-18-18.



Benefits of Orthophosphate

- Clean and clear with very few impurities
- Little to no precipitate formation in storage tank
- Low salt, can be put direct on the seed
- Phosphate is in ortho form and immediately plant available

- However, product is typically more expensive than other liquid phosphates.
- Must use EDTA micro's to add to mix.

Superphosphoric Acid

Commonly called polyphosphate, contains polymers which enables the solution to hold more P, and has the ability to hold its metal cations in solution when they would otherwise precipitate or settle out.

Typically an 85% acid solution, with an analysis of 0-68-0. Green in color and when reacted with water and ammonia to neutralize the acid, this makes polyphosphate products 10-34-0 and 11-37-0.



Is this what you look like when you make a new mixture?

